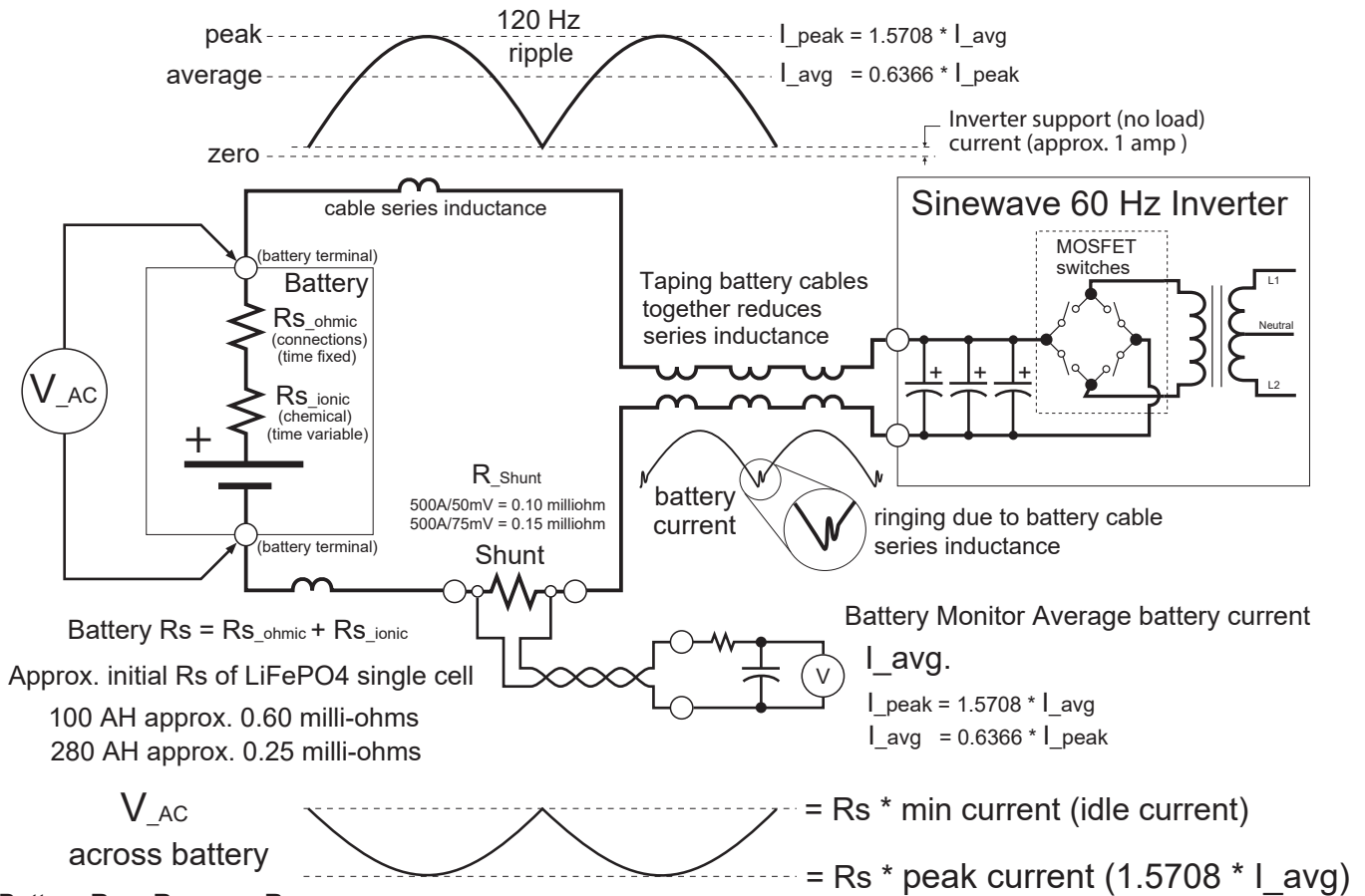


# DC current profile on battery

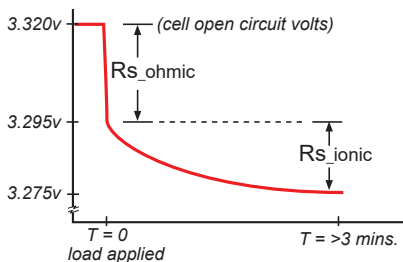
Battery peak current = 1.5708 x average current



Battery  $R_s = R_{s\_ohmic} + R_{s\_ionic}$ .

$R_{s\_ohmic}$  is material contact resistance and terminal resistance.  $R_{s\_ionic}$  is chemical ion movement resistance.  $R_{s\_ionic}$  is caused by localized depletion of available ions that have to be replaced with a transition movement delay. This results in exponential decay increasing apparent  $R_s$ , similar to long time constant RC discharge.  $R_{s\_ionic}$  looks like a resistance but it is actually chemical energy kinetics voltage drop required to transport ions. The greater the discharge current, the more kinetic voltage drop required to transport the needed quantity of ions. Eventually the  $R_{s\_ionic}$  resistance will nearly level out when equilibrium is reached at a given discharge current.  $R_{s\_ionic}$  resistance is a good indicator of battery quality. Battery impedance meters measure primarily  $R_{s\_ohmic}$ .  $R_s$  apparent of LiFePO4 single cell nearly doubles after 3 to 5 mins. of 0.5C discharge due to  $R_{s\_ionic}$  increase.  $R_{s\_ionic}$  resistance will recover to low current level within 5-15 minutes of load being removed. Measuring  $R_{s\_ionic}$  requires rapid current load pulsing as no load voltage will drop due to ion transport kinetic voltage drop.

0.5 C DC load on LiFePO4 cell



$$V_{AC \text{ pk-pk}} = R_s * (I_{pk-pk}), \quad I_{pk-pk} = 1.5708 * (I_{avg} - I_{idle}), \text{ assuming sinewave inverter}$$

$$V_{AC \text{ pk-pk}} = R_s * (1.5708 * I_{avg} - I_{idle})$$

$$R_s = V_{AC \text{ pk-pk}} / (1.5708 * (I_{avg} - I_{idle})), \quad V_{AC \text{ pk-pk}} = V_{rms} / 0.70711$$

$$R_s = (V_{rms} / 0.70711) / (1.5708 * (I_{avg} - I_{idle}))$$

**need true RMS AC coupled DVM to measure waveform**  
there will be an initial spike in DVM reading to charge its coupling cap.

Knowing the average DC current and AC voltage across battery the  $R_s$  of battery can be determined. (assumes sinewave inverter)

Example for 280 AH LiFePO4 cell with 50 - 100 amp load

- Prepare an inverter AC load that will draw 50-100 amps average DC battery current.
- Twist DVM leads to reduce stray pickup, keep leads at right angle to battery cables to reduce coupling.
- Connect AC true rms DVM to battery terminals directly (not battery cables on terminals).
- Measure inverter avg  $I_{DC}$  idle current from Batt Monitor, apply inverter load, measure  $I_{DC}$  and AC voltage after settles.

Measurements:  $V_{rms} = 27.8 \text{ mVrms}$ ,  $I_{idle} = 0.7 \text{ amps}$ ,  $I_{avg}$  with load = 97.2 amps

$$R_s = (V_{rms} / 0.70711) / (1.5708 * (I_{avg} - I_{idle})), \quad R_s = (27.8 \text{ mVrms} / 0.70711) / (1.5708 * (97.2 \text{ A} - 0.7 \text{ A}))$$

**$R_s = 0.259 \text{ milli-ohms}$**